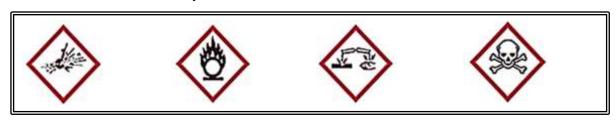
## YEAR 13 CHEMISTRY WORSHEET 1 - 2021

## **Strand 1: General Chemistry**

1. Name the chemical hazard symbols shown below.



- (i) \_\_\_\_\_ (ii) \_\_\_\_ (iv) \_\_\_\_
- 2. Describe how leftover **harmless soluble** salts such as CaCl<sub>2</sub> and MgSO<sub>4</sub> should be disposed.

- 3. **Product X** can be prepared by **process A** and **process B**. **Process A** has an atom economy of **30 %** whereas **process B** has an atom economy of **80 %**.
  - (i) Identify the **process** which is greener.

\_\_\_\_\_

(ii) Provide a reason for your answer to part (i) above.

4. Calculate the atom economy for the preparation of  $NH_3$  from  $(NH_4)_2SO_4$ .

$$(NH_4)_2SO_{4(s)} \longrightarrow H_2SO_{4(aq)} + 2NH_{3(aq)}$$

 $[M (H_2SO_4) = 98; M (NH_3) = 17]$ 

